SAMBA ITT

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Dynamics of Fine Bubbles

Transport Mechanisms



Garcia-Ochoa et al. (2010) Biochemical Engineering Journal 49 pp. 289–307

Governing Equations

Generalised Rayleigh-Plesset (R-P)
$$\frac{p_B(t) - p_{\infty}(t)}{\rho_L} = R \frac{d^2 R}{dt^2} + \frac{3}{2} \left(\frac{dR}{dt}\right)^2 + \frac{4\nu_L}{R} \frac{dR}{dt} + \frac{2S}{\rho_L R}$$

With impurities, R-P becomes

$$\frac{p_{vap}(T_{\infty}) - p_{\infty}(t)}{\rho_L} + \frac{p_{vap}(T_B) - p_{\infty}(t)}{\rho_L} + \frac{p_{Go}}{\rho_L} \left(\frac{T_B}{T_{\infty}}\right) \left(\frac{R_o}{R}\right)^3$$
$$= R \frac{d^2 R}{dt^2} + \frac{3}{2} \left(\frac{dR}{dt}\right)^2 + \frac{4\nu_L}{R} \frac{dR}{dt} + \frac{2S}{\rho_L R}$$

Population balance – multiple size group $\frac{\partial \rho_G \varepsilon_G f_i}{\partial t} + \nabla \cdot (\mathbf{u}_G \rho_G \varepsilon_G f_i) = \rho_G S_i \quad i = 1, 2 \dots M$

Transport mechanism $\frac{1}{K_L a} = \frac{1}{H \cdot k_G a} + \frac{1}{E \cdot k_L a}$ Reaction kinetics Coalescence Break-up Turbulence Non-Newtonian

Dynamics of Fine Bubbles

Key Questions to Answer



- Use R-P equation to describe the growth and break-up of microbubbles
- Population balance to model bubble size distribution
- Main resistance to mass transfer

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Bubble-less Ozonation





Ozone – powerful oxidant

- Eliminate micro-pollutants from wastewater
- Pollutant degradation through oxidation
- Reacts with a large number of compounds
- No storage or handling of chemicals



Bubble-less Ozonation

Membrane Contactor



Transport & Reaction Kinetics



Bubble-less Ozonation

Key Questions to Answer



- How reaction kinetics affect concentration profiles?
- Most suitable membrane –

geometry and type

Correlations for prediction

O. Levenspiel (1999) Chemical Reaction Engineering. Wiley & Sons.

Extremely fast reaction