

Pharmaceutical Surface Science: Probing at the Nanometre Scale

Applications to drug-polymer interactions in inhalation systems



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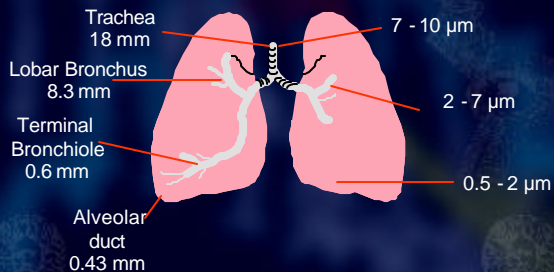
Synopsis

- Introduction
- AFM Studies
 - Drug-polymer interactions in DPI formulations.
 - Drug-polymer interactions in model suspension pMDI systems.
- Conclusions
- Acknowledgments

Targeting by deposition

Airway diameter

Deposition particle diameter



Dry Powder Inhalers (DPIs)

Key issues:

- Drug Particles must reach the deep lung.
- Particle size must be below 5µm.
- Powders must be micronised.
- Micronised powders are very cohesive

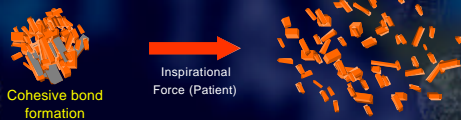
Currently less than 20% of the emitted dose reaches the deep lung!

Dry Powder Aerosols: Formulation Strategies

Carrier - Based Systems



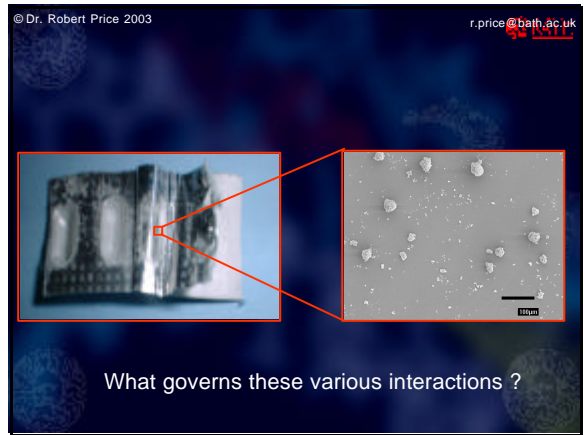
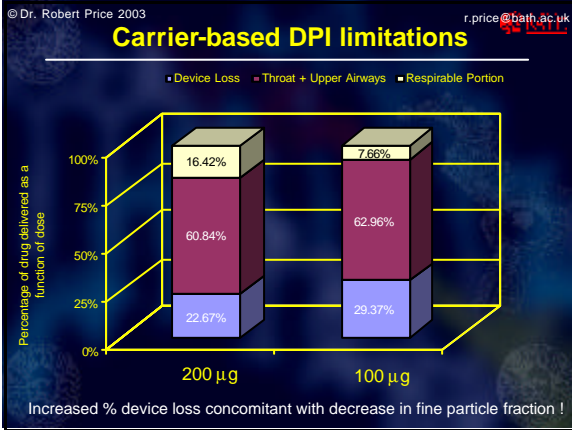
Agglomerated Systems



Interactions to be controlled and modified

- Drug - Drug Interactions (Cohesion)
- Drug - Excipient Interactions (Adhesion)
- Drug - Device Interactions (Segregation)

Their properties govern overall stability and aerosol delivery performance of a formulation



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A composite of Interparticulate Forces

Particle interactions are primarily dictated by:

- van der Waals Forces
- Electrostatic Forces
- Capillary Forces

The relative contribution of these components to the total adhesion/cohesion depends on the interacting materials and relative humidity.

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Physico-chemical Link

- Lifshitz (1955) grouped the electrodynamic interactions:
 - London-van der Waals
 - Keesom-van der Waals
 - Debye-van der Waals
 Collectively known as Lifshitz-van der Waals (LW) interactions
- Electron-donor and electron acceptor interactions are AB interactions, for Lewis acid-base.

$$\gamma_s^T = \gamma_s^{LW} + 2\sqrt{\gamma_s^+ \gamma_s^-}$$

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Work of adhesion (W_a)

$$W_a = \gamma_1 + \gamma_2 - \gamma_{12}$$

$$W_a = 2\left(\sqrt{\gamma_1^{LW} \gamma_2^{LW}} + \sqrt{\gamma_1^+ \gamma_2^-} + \sqrt{\gamma_1^- \gamma_2^+}\right)$$

LW attraction between solid 1 and 2 Polar adhesion between solid 1 and 2

How is the W_{ad} related to a measured interfacial force ?

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Pull-off Forces

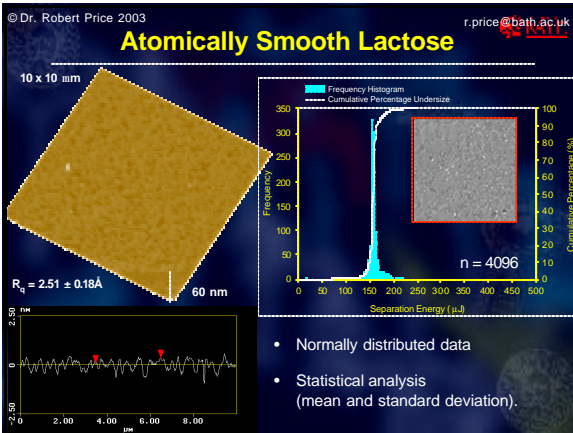
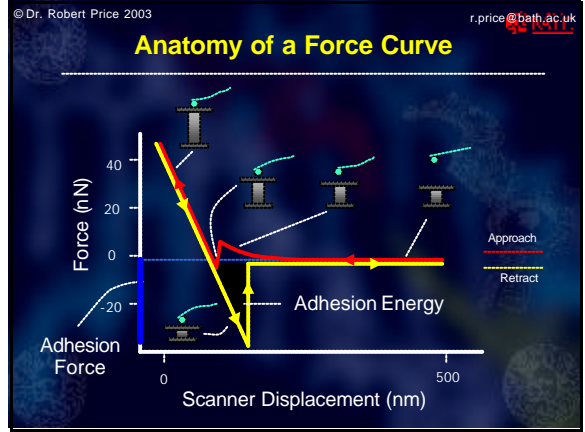
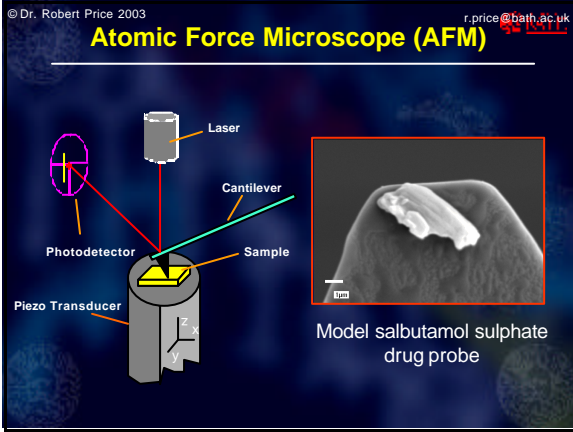
$$F_{pull-off} = 2\pi R W_{ad} \text{ (DMT)}$$

$$F_{pull-off} = 3/2\pi R W_{ad} \text{ (JKR)}$$

$$F_{pull-off} = 4\pi R \left(\sqrt{\gamma_1^{LW} \gamma_2^{LW}} + \sqrt{\gamma_1^+ \gamma_2^-} + \sqrt{\gamma_1^- \gamma_2^+} \right)$$

$$F_{pull-off} = 3\pi R \left(\sqrt{\gamma_1^{LW} \gamma_2^{LW}} + \sqrt{\gamma_1^+ \gamma_2^-} + \sqrt{\gamma_1^- \gamma_2^+} \right)$$

$F_{ad} \propto W_{ad}$

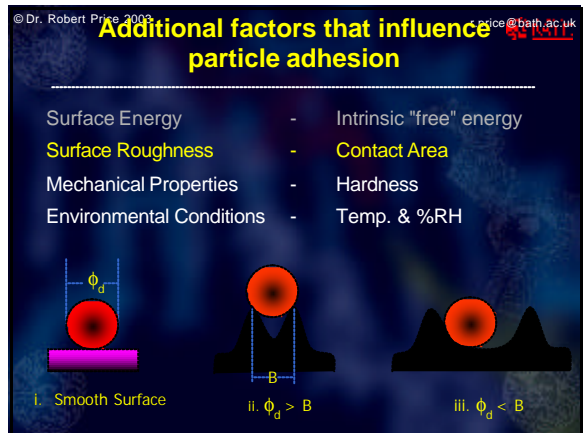
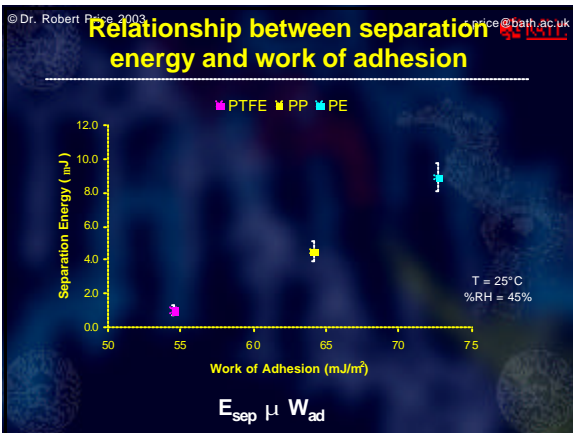


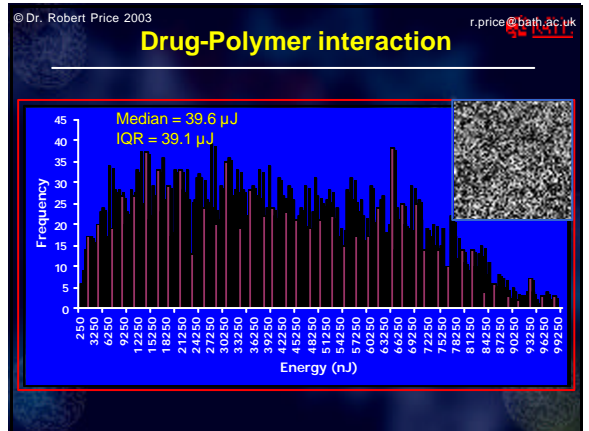
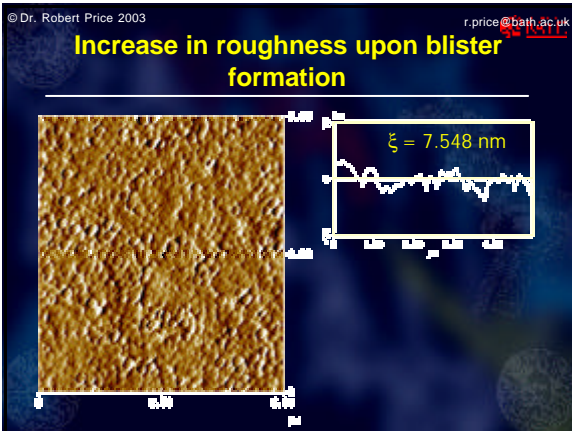
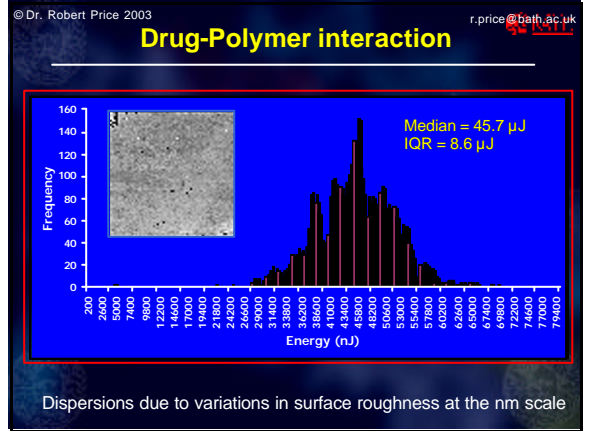
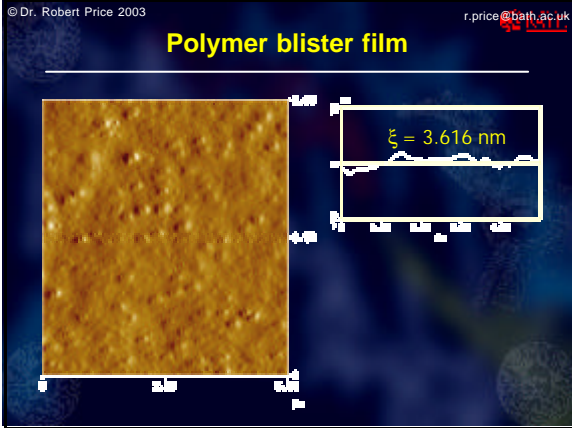
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Energetic analysis of experimental measurements

Material	γ^W (mJ/m ²)	γ^* (mJ/m ²)	γ (mJ/m ²)	W_a (mJ/m ²)
Salbutamol*	41.5	19.3	5.8	126.52
Lactose*	47.90	28.00	5.70	135.64
PTFE	18.5	0	0	54.41
PP	25.7	0	0	64.12
PE	33.0	0	0	72.66
PVC	43.0	0.04	3.5	85.32

*D. Cline and R. Dalby, Pharm Res. 19 (2002) 1274-1277

$$W_a = 2\sqrt{\gamma_1^W \gamma_2^W} + \sqrt{\gamma_1 \gamma_2} + \sqrt{\gamma_1 \gamma_2}$$




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- ## Summary
- Surface energy properties of substrate surface play a dominant role in the adhesion of individual drug particles.
 - Adhesion of micron sized particles is highly dependant on the surface roughness at the nanometre scale.
 - The presence of additives may lead to instability and a source of segregation of drug particulates in a dry powder formulation.

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pMDI Model Interactions

$\gamma_3^{LW}, \gamma_3^+, \gamma_3^-$

$\gamma_1^{LW}, \gamma_1^+, \gamma_1^-$

$\gamma_2^{LW}, \gamma_2^+, \gamma_2^-$

$\gamma_3^+, \gamma_3^- = 0$

$$W_{132} = 2 \left(\sqrt{\gamma_1^{LW} \gamma_3^{LW}} + \sqrt{\gamma_2^{LW} \gamma_3^{LW}} - \sqrt{\gamma_1^{LW} \gamma_2^{LW}} - \gamma_3^{LW} \right) - \sqrt{\gamma_1^+ \gamma_2^-} - \sqrt{\gamma_1^- \gamma_2^+}$$

LW repulsion between solid 1 and solid 2 with solution

LW attraction between solid 1 and solid 2
LW attraction of solution

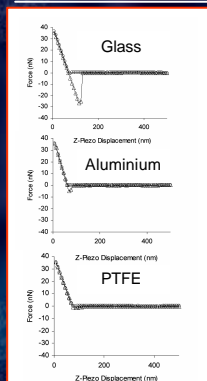
Polar adhesion between solid 1 and 2

Drug-Canister model interactions

Material (2)	Solvent (3)	Repulsive LW interaction (mJ/m ²)	Attractive LW interaction (mJ/m ²)	Attractive AB interaction (mJ/m ²)	W ₁₃₂ Total interaction (mJ/m ²)
Borosilicate Glass	HPFP	104.22	-126.41	-9.21	31.40
Anodised Aluminium	HPFP	96.81	-111.02	-4.36	17.67
PTFE	HPFP	89.32	-95.46	-1.65	7.80

Drug particle (1): Salbutamol sulphate (micronised)
 Model Propellant: 2H, 3H perfluoropentane (HPFP)

Direct Force Measurements

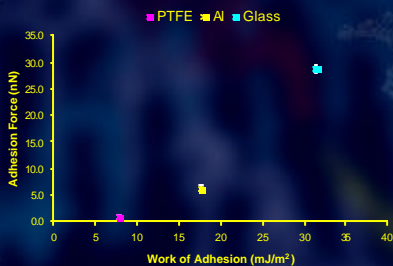


$F_{ad} = 28.63 \text{ nN}$ ($W_{132} = 31.40 \text{ mJ/m}^2$)

$F_{ad} = 6.24 \text{ nN}$ ($W_{132} = 17.67 \text{ mJ/m}^2$)

$F_{ad} = 0.85 \text{ nN}$ ($W_{132} = 7.80 \text{ mJ/m}^2$)

pMDI drug-wall interactions



$F_{ad} \propto W_{ad}$

Conclusions

- AFM can be used to determine the adhesive characteristics of particulate materials to polymer surfaces.
- In combination with bulk techniques, AFM may potentially play a pivotal role in the design and modifications of DPI and suspension based systems.
- There is a further need to correlate:
 - Relationship between surface thermodynamics (contact angles, IGC), force measurements (AFM, CPD) and *in vitro* performance.
 - Macroscopic properties of surfaces and meso scale properties of interfacial interactions and related adhesion.

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100µm