

## RITZ VARIATION METHOD

Series of trial  $\psi$ 's       $\psi_1, \psi_2, \dots$  corresponding energies  $E_1, E_2, \dots$  calculated.

Minimum  $E$  corresponds to the best  $\psi$

Actual  $E < E_{\min}$  found

### Special case of the above - LINEAR COMBINATIONS METHOD

Put  $\psi = c_1\psi_1 + c_2\psi_2 + c_3\psi_3 + \dots + c_n\psi_n$

Variation method applied to obtain values of  $c$  which give *minimum energy* for  $\psi$

1. With  $n$  terms in the expression for  $\psi$ , the equation for  $E$  is of the  $n^{\text{th}}$  order giving  $n$  values: ground state and excited states.
2. The importance of the contribution of  $\psi_1, \psi_2, \psi_3$  etc. to the actual wavefunction is  $c_1: c_2: c_3: \dots$  etc.
3. The greater the overlap between  $\psi_1, \psi_2$  etc. the lower the energy of the corresponding  $\psi$ . Thus high overlap leads to a strong bond.

### Molecular orbitals: LINEAR COMBINATION OF ATOMIC ORBITALS (L.C.A.O.)

$$\psi = c_A\psi_A + c_B\psi_B$$

$\psi_A$  and  $\psi_B$  are for **atomic orbitals**

## APPLICATION OF THE VARIATION METHOD TO ONE-DIMENSIONAL WELL

An exact solution is straight forward:

$$\psi = (2/L)^{1/2} \sin n\pi x/L$$

but, for sake of illustration, consider trial solutions.

A cubic (over part of its range) might be suitable. One which fits the boundary conditions is

$$\psi_1 = ax^3 - aLx^2$$

Applying the normalisation condition gives

$$a^2 = 105/L^7$$

Using the energy function to evaluate the energy gives

$$E = 7\hbar^2/mL^2$$

A *better solution* might be found using the method of linear combinations. If  $\psi_1$  is a possible solution, the complementary cubic could also be

$$\psi_2 = -ax^3 + 2aLx^2 - aL^2x$$

By symmetry

normalisation condition

$$a^2 = 105/L^7$$

energy

$$E = 7\hbar^2/mL^2$$

### Linear combination

A linear combination should give a better wavefunction:

$$\psi_3 = c_1\psi_1 + c_2\psi_2$$

Applying the variation method gives

$$c_1/c_2 = \pm 1$$

(ground state and excited state)

The new wavefunction is much better, i.e. the energy predicted is much closer to that of the exact solution.

	Relative energy $E \times mL^2/\hbar^2$
<b>1st excited state</b>	
Linear combination of cubics	21
Exact	19.6
<b>Ground state</b>	
Single cubic	7
Linear combination of cubics	5
Exact	4.9